

<b>Scales E3 Mark Scheme</b>		
<b>1(a)</b>	3 cm	[1]
<b>1(b)</b>	1 cm	[1]
<b>1(c)</b>	2 cm	[1]
<b>1(d)</b>	2 cm	[1]
<b>1(e)</b>	1 cm	[1]
<b>2(a)</b>	2.5 cm	[1]
<b>2(b)</b>	0.9 cm	[1]
<b>2(c)</b>	3.6 cm	[1]
<b>2(d)</b>	1.4 cm	[1]
<b>2(e)</b>	6.2 cm	[1]
<b>3(a)</b>	4 cm	[1]
<b>3(b)</b>	2 cm	[1]
<b>3(c)</b>	1 cm	[1]
<b>4(a)</b>	10 cm	[1]
<b>4(b)</b>	$10 \div 10 = 1$ cm	[1]
<b>5(a)</b>	20°C	[1]
<b>5(b)</b>	70°C	[1]
<b>5(c)</b>	40°C	[1]
<b>6(a)</b>	60°F	[1]
<b>6(b)</b>	90°F	[1]
<b>6(c)</b>	150°F	[1]
<b>7(a)</b>	44°C	[1] Accept 43 – 45
<b>7(b)</b>	67°C	[1] Accept 66 – 68
<b>7(c)</b>	7°C	[1] Accept 6 – 8

<b>8(a)</b>	44°F	[1] Accept 43 – 45
<b>8(b)</b>	55°F	[1] Accept 54 – 56
<b>8(c)</b>	114°F	[1] Accept 113 – 115
<b>9(a)</b>	67°C	[1] Accept 66 – 68
	153°F	[1] Accept 152 – 154
<b>9(b)</b>	40°C	[1] Accept 39 – 41
	104°F	[1] Accept 103 – 150
<b>9(c)</b>	50°C	[1] Accept 49 – 51
	122°F	[1] Accept 121 – 123
<b>10(a)</b>	7°C	[1] Accept 6 – 8
<b>10(b)</b>	14°C	[1] Accept 13 – 15
	14 – 7 = 7°C warmer	[1] Accept ft with their readings
<b>11(a)</b>	40°C	[1]
	105°F	[1] Allow answer in range 104 – 106
<b>11(b)(i)</b>	105 – 18 = 87°F	[1] ] Allow answer in range 86 – 88
<b>11(b)(ii)</b>	30°C or 31°C (from scale on thermometer)	[1] Allow 28°C or 29°C
<b>11(b)(iii)</b>	40 – 30 = 10°C or 40 – 31 = 9°C	[1] Allow 12°C or 11°C
<b>12(a)</b>	600 ml	[1]
<b>12(b)</b>	100 ml	[1]
<b>12(c)</b>	1000 ml	[1]
<b>13(a)</b>	75 ml	[1]
<b>13(b)</b>	825 ml	[1]
<b>13(c)</b>	550 ml	[1]

<b>14(a)</b>	600 ml in left	[1]
	300 ml in right	[1]
<b>14(b)</b>	$600 \div 2 (= 300)$	[1]
	$300 + 300 (= 600)$	[1]
	300 ml in left, 600 ml in right	[1]
<b>14(c)</b>	$600 \div 3 (= 200)$	[1] Allow ecf from (b)
	$300 + 200 (= 500)$	[1] Allow ecf from (b)
	500 ml in left, 400 ml in right	[1] Allow ecf from (b)

<b>15(a)</b>	900 g	[1]
<b>15(b)</b>	300 g	[1]
<b>15(c)</b>	600 g	[1]

<b>16(a)</b>	700 g	[1]
<b>16(b)</b>	1000 g	[1]
<b>16(c)</b>	100 g	[1]

<b>17(a)</b>	6 oz	[1]
<b>17(b)</b>	22 oz	[1]
<b>17(c)</b>	14 oz	[1]
<b>17(d)</b>	(a) and (c) are lighter, (b) is heavier	[1]

<b>18(a)</b>	Left weighs 50 g	[1]
	Right weighs 150 g	[1]
	$150 - 50 = 100$ g	[1]
<b>18(b)</b>	Left	[1]
	Because it is lighter	[1]
<b>18(c)</b>	$150 - 125 = 25$ g	[1]

<b>19(a)</b>	300 ml	[1]
	600 g	[1]
<b>19(b)</b>	$600 \div 300 (= 2)$	[1]
	2 g	[1]
<b>19(c)</b>	$300 + 300 = 600 \text{ ml}$	[1]
	$600 \div 2 (= 300)$	[1]
	$600 + 300 = 900 \text{ g}$	[1]